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Microbial production of vitamin C

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Microbial production of vitamin C

The present invention relates to the microbial production of L-ascorbic acid (vitamin C).

5 Vitamin C, which is one of very important and indispensable nutrient factors for human beings, has been commercially produced by the so-called "Reichstein method", which is well known as a technologically established process. This method, however, comprises a number of complex steps and any improvement in the overall yield is difficult to achieve. Therefore, there have been a number of proposals, which contemplate a reduction in the number of steps and/or an improvement in the overall yield.

10 The present invention provides a process for the production of vitamin C from L-sorbose which comprises contacting a microorganism belonging to the genus *Ketogulonicigenium* with L-sorbose in a reaction mixture and isolating and purifying vitamin C from the reaction mixture.

15 As used herein, "contacting a microorganism with L-sorbose in a reaction mixture" includes cultivation of the microorganism in a medium containing L-sorbose. The microorganism may be further used, for instance, in the form of resting cells, acetone treated cells, lyophilized cells, immobilized cells and the like to act directly on the substrate, i.e. L-sorbose. Any means per se known as a method in connection with the incubation technique for microorganisms may be adopted through the use of aeration,
20 preferably agitated submerged fermenters. The preferred cell concentration range for carrying out the reaction is from about 10 mg to about 700 mg of wet cell per ml, more preferably from about 30 mg to about 500 mg of wet cell per ml.

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- 2 -

A suitable "reaction mixture" could be water or any nutrient medium known in the art for the cultivation of the microorganism. Such nutrient medium includes a carbon source, a nitrogen source and other inorganic salts, which can be utilized by the microorganism. Various nutrient materials which are generally used for the better growth of
5 microorganisms may suitably be included in the medium.

Examples of such nutrients as assimilable carbon sources include, but are not limited to, glycerol, D-mannitol, erythritol, ribitol, xylitol, arabinol, inositol, dulcitol, D-ribose, D-fructose, D-glucose and sucrose. Examples of digestible nitrogen sources such as organic substances, include, but are not limited to, peptone, yeast extract, baker's yeast, urea,
10 amino acids and corn steep liquor. Various inorganic substances may also be used as nitrogen sources, for example nitrates and ammonium salts. Furthermore, the culture medium usually contains inorganic salts, for example, magnesium sulfate, potassium phosphate and calcium carbonate.

In one embodiment, the present invention provides a process for the production of
15 vitamin C from L-sorbose which comprises contacting a microorganism which is selected from *Ketogulonicigenium robustum*, *Ketogulonicigenium vulgare* or mutants thereof with L-sorbose in a reaction mixture and isolating and purifying vitamin C from the reaction mixture.

As used herein, the microorganisms "*Ketogulonicigenium robustum*" or "*Ketogulonigenium robustum*" and "*Ketogulonicigenium vulgare*" or "*Ketogulonigenium vulgare*" also include
20 synonyms or basonyms of such species having the same physiological properties, as defined by the International Code of Nomenclature of Prokaryotes.

As used herein, "mutants" of the microorganisms mentioned above refer to microorganisms which are altered in their genomic sequences which are capable of the
25 conversion of L-sorbose to vitamin C as provided by the process of the present invention. Mutants may be obtained by any convenient means including, for example, chemical and UV mutagenesis, followed by screening or selection for a desired phenotype, construction of dysfunctional genes *in vitro* by recombinant techniques used to replace the intact counterparts of the genes in the genome of the microorganism, by single and double
30 cross-over recombinations, and other well known techniques. See, Sambrook, *et al.*, Molecular Cloning, A Laboratory Manual, 2nd Ed., Cold Spring Harbor Laboratory Press (1989) and, Harwood and Cutting, Molecular Biology Methods For *Bacillus*, John Wiley and Sons (1990), pp. 27-74. Suitable mutagens include, but are not limited to, ultraviolet-ray, X-ray, γ -ray and chemical mutagens such as nitrogen mustard or N-methyl-N'-nitro-
35 N-nitrosoguanidine. Furthermore, a mutant can be obtained by isolating a clone

- 3 -

occurring by spontaneous mutation thereof in any of the ways per se well known for the purpose by one skilled in the art.

In a preferred embodiment, the microorganisms as used for the process of the present invention are selected from the group consisting of *K. robustum* NRRL B-21627, *K. vulgare* NRRL B-30035, *K. vulgare* NRRL B-30036 and *K. vulgare* NRRL B-30037N.

K. robustum NRRL B-21627 is described in US 5,834,231. Strains *K. vulgare* NRRL B-30035, *K. vulgare* NRRL B-30036 and *K. vulgare* NRRL B-30037N are described in US 6,316,231 B1. These strains are publicly available from the American Type Culture Collection, 1815 N. University Street, Peoria, Illinois 61604, USA.

Cultivation of the microorganisms of the present invention may be conducted at a pH of about 4.0 to about 9.0, wherein a pH of about 5.0 to about 8.0 may preferably be maintained. The cultivation period varies depending on the pH, temperature and nutrient medium to be used, and is preferably about 1 to 5 days, most preferably about 1 to 3 days. The preferred temperature for carrying out the process of the present invention is a temperature of about 13 to about 36°C, more preferably of about 18 to about 33°C.

In one embodiment the process of the present invention is carried out at a pH of about 4.0 to about 9.0 and at a temperature of about 13 to about 36°C. Preferably, a pH of about 5.0 to about 8.0 and a temperature of about 18 to about 33°C is used for carrying out the inventive process.

Although the concentration of L-sorbose may vary with the reaction conditions, the reaction is preferably carried out at L-sorbose concentrations of about 2 to about 120 mg/ml, more preferably at concentrations of about 4 to about 100 mg/ml.

The vitamin C thus produced and accumulated in the reaction mixture may be separated and purified by any per se known conventional method which suitably utilized the property of the product, and it may be separated as the free acid or as a salt of sodium, potassium, calcium, ammonium or the like.

Specifically, the separation may be performed by any suitable combination or repetition of the following steps: by the formation of a salt, by using differences in properties between the product and the surrounding impurities, such as solubility, absorbability and distribution coefficient between the solvents, by absorption, for example on ion exchange resin. Any of these procedures alone or in combination constitute a convenient means for isolating the product. The product thus obtained may further be purified in a conventional manner, for example, by re-crystallization or chromatography.

- 4 -

According to the present invention, the improvement in terms of the reduction in the number of steps is very significant because it leads to one step pathway directed to the production of the vitamin C from L-sorbose.

In the following Example, the process of the present invention will be illustrated in more detail.

Example 1: Production of vitamin C from L-sorbose with resting cell system

K. robustum NRRL B-21627 and K. vulgare strains NRRL B-30035, NRRL B-30036 and NRRL B-30037N were cultivated on Tryptic Soy Agar (Difco, Becton, Dickinson and Company, Sparks, MD, USA) at 30°C for 3 days. The cells were harvested from the plate and suspended into 1 ml of 50 mM potassium phosphate buffer (pH 7.0) and washed twice with the same buffer. The optical density of the cell suspensions at 600 nm were 12.2, 12.5, 16.2 and 11.2 for strains NRRL B-21627, NRRL B-30035, NRRL B-30036 and NRRL B-30037, respectively. These numbers corresponded to 31.7, 32.5, 42.1 and 29.1 mg of wet cell weight per ml, respectively.

The reaction mixture (5 ml in test tube) contained 0.9 ml of the cell suspension and 0.1 ml of 50 mg/ml L-sorbose in 50 mM potassium phosphate buffer (pH 7.0). The reaction was started by the addition of the cell suspension. The reaction mixture was incubated at 30°C and with 180 rpm on a reciprocal shaker for 3 hours. After the reaction, the reaction mixture was centrifuged at 8,000 x g for 10 min to obtain the supernatant. The vitamin C content in the supernatant was measured with HPLC:

Column: YMC-Pack Polyamine II (150 x 4.6 mm i.d.), YMC Co.Ltd, Kyoto, Japan, Eluent: 50 mM NH₄H₂PO₄/Acetonitrile=30/70

Flow rate: 1 ml/min,

Detection: UV absorption at 250 nm

The retention time of vitamin C under the HPLC conditions above was 7.7 min. The reaction product with all the strains tested was confirmed to be vitamin C with this HPLC analysis.

Table 1 shows the quantity of vitamin C produced by strains NRRL B-21627, NRRL B-30035, NRRL B-30036 and NRRL B-30037N.

- 5 -

Table 1: Vitamin C production from L-sorbose

Strain	Vitamin C produced (mg/ml)
<i>K. robustum</i> NRRL B-21627	0.21
<i>K. vulgare</i> NRRL B-30035	0.44
<i>K. vulgare</i> NRRL B-30036	0.36
<i>K. vulgare</i> NRRL B-30037N	0.46

- 6 -

Claims

1. A process for the production of vitamin C from L-sorbose which comprises contacting a microorganism belonging to the genus *Ketogulonigenium* with L-sorbose in a reaction mixture and isolating and purifying vitamin C from the reaction mixture.
- 5 2. The process according to claim 1, wherein the microorganism is selected from *Ketogulonigenium robustum*, *Ketogulonigenium vulgare*, or mutants thereof.
3. The process according to claim 1 or 2, wherein the microorganism is selected from the group consisting of *Ketogulonigenium robustum* NRRL B-21627, *Ketogulonigenium vulgare* NRRL B-30035, *Ketogulonigenium vulgare* NRRL B-30036 and *Ketogulonigenium*
10 *vulgare* NRRL B-30037.
4. The process according to any one of claims 1 to 3, wherein the process is carried out at a pH of about 4.0 to about 9.0 and at a temperature of about 13 to about 36°C.
5. The process according to any one of claims 1 to 4, wherein the process is carried out at a pH of about 5.0 to about 8.0 and at a temperature of about 18 to about 33°C.
- 15 6. The process according to any one of claims 1 to 5, wherein the process is carried out at a L-sorbose concentration of about 2 to about 120 mg/ml.
7. The process according to claim 6, wherein the process is carried out at a L-sorbose concentration of about 4 to about 100 mg/ml.
